Centre No.			Paper Reference				Surname	Initial(s)			
Candidate No.			6	P	H	0	5	/	1	Signature	

Paper Reference(s)

6PH05/1 Edexcel GCE Physics

Exam	Examiner's use only								
Team L	Team Leader's use only								

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19

Total

Unit 5: Physics from Creation to Collapse

Sample Assessment Material

Time: 1 hour 35 minutes

| Materials required for examination | Nil | Items included with question papers | Nil |

Instructions to Candidates

In the boxes above, write your centre number, candidate number, your surname, initial(s) and signature. Check that you have the correct question paper.

Answer ALL the questions in Section A and Section B. Write your answers in the spaces provided in this question paper.

Some questions must be answered with a cross in a box (\boxtimes). If you change your mind, put a line through the box (\boxtimes) and then mark your new answer with a cross (\boxtimes).

Information for Candidates

The marks for individual questions and the parts of questions are shown in round brackets: e.g. (4). The total mark for this paper is 80. There are 19 questions in this paper. There are 20 pages in this question paper. Any blank pages are indicated.

Advice to Candidates

Quality of written communication will be taken into account in the marking of your responses to Questions 18 and 19. These questions are indicated with an asterisk. Quality of written communication includes clarity of expression, the structure and presentation of ideas and grammar, punctuation and spelling.

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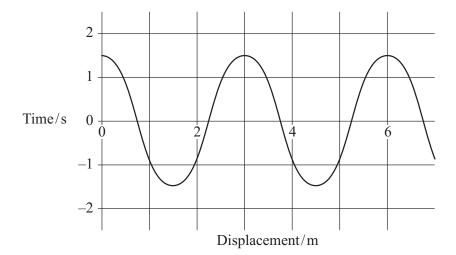
Turn over



chang . A va	SECTION A Answer ALL the questions in this section. It is a cross in the box (⋈). It is a cross in the box (⋈). Answer ALL the questions in this section.		
chang . A va	ach question, select one answer from A to D and put a cross in the box (☒). It nge your mind, put a line through the box (☒) and then mark your new answer with a cross (☒).		
	valid set of units for specific heat capacity is		
×			
	$A kg J^{-1} K^{-1}$		
\times	$\mathbf{B} = \mathrm{kg}\mathrm{J}\mathrm{K}^{-1}$		
×	$C = kg^{-1} J K^{-1}$		
×	\mathbf{D} kg $\mathbf{J}^{-1}\mathbf{K}$		Q1
	(Total 1 n	nark)	
×			
	C g		
X	$\mathbf{D} = g/4$		Q2
	(Total 1 n	nark)	

Leave blank

3. A child is playing on a swing. The graph shows how the displacement of the child varies with time.



The maximum velocity, in m s⁻¹, of the child is

- $\mathbf{A} \quad \pi/2$
- \boxtimes **B** π
- \square C 2π
- \square **D** 3π

Q3

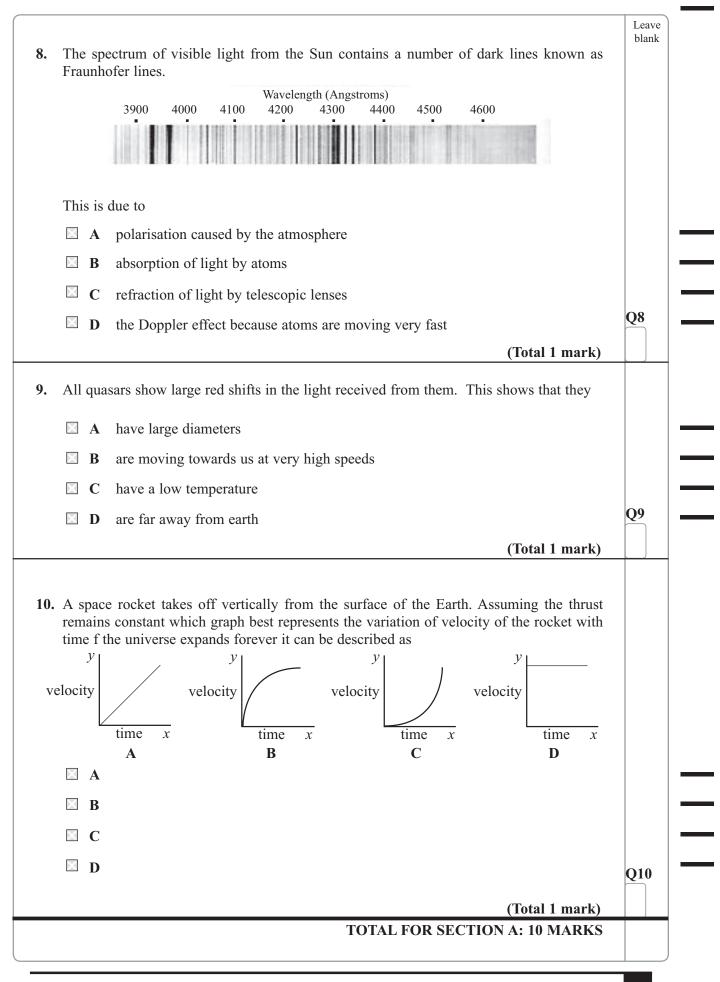
(Total 1 mark)

- **4.** Cosmic background radiation is a remnant of the big bang and appears to pervade the universe. It has a maximum wavelength in the microwave region of the electromagnetic spectrum. This can be calculated to correspond to a temperature of about 3 K. This calculation is based on the assumption that
 - \square **A** the universe is spherical
 - B the universe is expanding
 - C space can be regarded as a black body
 - **D** space is a vacuum

Q4

(Total 1 mark)

5.	She	e wri	ent is asked: "What is meant by background radiation?" ites: "It is the radiation produced by the rocks in the ground." swer is	
	X	A	correct	
	X	В	incorrect – rocks do not produce radiation	
	X	C	incomplete	
	X	D	incorrect – this radiation would not be ionising	
			(Total 1 mark)	
6.			river notices that her rear view mirror shakes a lot at a particular speed. To try to she sticks a big lump of chewing gum on the back of the mirror.	
	Wh	nich	one of the following statements is correct?	
	X	A	The mirror no longer shakes a lot because it is heavily damped.	
	X	В	The mirror stills shakes a lot at the same speed as before because the chewing gum does not change the damping.	
	X	C	The mirror shakes a lot at a different speed because the chewing gum changes the damping.	
	X	D	The mirror shakes a lot at a different speed because the chewing gum has changed the resonant frequency of the mirror.	
			(Total 1 mark)	
7.	The	e <i>x</i> -a	xis of a Hertzsprung-Russell diagram is $\log T$. This is because	
	X	A	the range of temperatures of the surfaces of stars is large	
	X	В	the temperatures of the surfaces of stars are all very large numbers	
	X	C	the scale has to start with the hottest stars	
	X	D	the diagram would be impossible to interpret if $\log T$ was the y-axis	
			(Total 1 mark)	



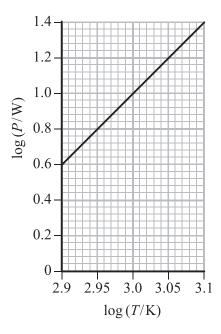
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n this section.	
3 m 3 .	
as is only kinetic.	
	(2)
es temperature is 27 °C. Caner.	alculate the
olecules =	(2) Q11

SECTION B		
Answer ALL the questions in this section.		
11. An ideal gas is contained in a volume of $2.0 \times 10^{-3} \mathrm{m}^3$.		
(a) Explain why the internal energy of an ideal gas is only kinetic.		
	••••	
	••••	
	(2)	
(b) The pressure of the gas is 1.2×10^5 Pa and its temperature is 27 °C. Calculate number of molecules of gas within this container.	the	
Number of molecules =		
	(2)	Q11
(Total 4 mar	ks)	

(4)	Describe how you would determine whether this radioactive source emits alpha particles only.
<i>a</i>)	(4)
(b)	State why smoke detectors do not provide a radiation risk in normal use.
	(1)
	(Total 5 marks)
tho wh	o stars in the night sky appear equally bright to an observer. The Ancient Greeks ught that all stars were the same distance from the Earth. State and explain two reasons y these two stars do not need to be the same distance from the observer.
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Leave blank

14. The graph shows how the logarithm of the electrical power P supplied to a filament lamp varies with the logarithm of the temperature T of the filament.



(a) P is related to T by a power law: $P = k T^n$ Use the graph to determine n.

n =(2)

(b) A student suggests that this relationship is predicted by the Stefan-Boltzmann law.

Comment on this statement.

(Total 4 marks)

(2)

Q14

15. The	e Hubble constant is thought to be about 70 000 m s ⁻¹ Mpc ⁻¹ .	Leave blank
	Give one reason why the value of this constant is uncertain.	
	(1)	
(b)	State how an estimate of the age of the Universe can be calculated from the Hubble constant.	
	(1)	
(c)	Explain how the ultimate fate of the Universe is associated with the Hubble constant.	
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(c)	Explain how the ultimate fate of the Universe is associated with the Hubble	Q15

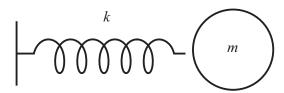
Pot	assium-40 ($_{19}$ K) is unstable.	
(a)	Calculate the binding energy per nucleon for potassium-40. Nuclear mass of potassium-40 = 39.953548 u Mass of one neutron = 1.008 665 u Mass of one proton = 1.007 276 u	
	Binding energy per nucleon =	
		(6)

(1)

(c)	Scientists have worked out the age of the Moon by dating rocks brought back by the	ne
	Apollo missions. They use the decay of potassium-40 to argon-40. The half-life optassium-40 is 1.3×10^9 years.	
	(i) Show that the decay constant of potassium-40 is about $5 \times 10^{-10} y^{-1}$.	
		••
		••
	(1)
	(ii) In one rock sample the scientists found $0.84\mu g$ of argon-40 and $0.10\mu g$ potassium-40.	of
	Calculate the age of the rock sample in years.	
	Age of rock =	
		4)
	(Total 12 marks	c)

Leave	
blank	

17. Certain molecules such as hydrogen chloride (HCl) can vibrate by compressing and extending the bond between atoms. A simplified model ignores the vibration of the chlorine atom and just considers the hydrogen atom as a mass m on a spring of stiffness k which is fixed at the other end.



(a) (i) Show that the acceleration of the hydrogen atom, a, is given by $a = -\frac{kx}{m}$ where x is the displacement of the hydrogen atom.

•••••	 			
• • • • • • • • • • • • • • • • • • • •	 	•••••	•••••	
				(2)

(ii) Hence derive the equation $T = 2\pi \sqrt{\frac{m}{k}}$ for the period of natural oscillations of the hydrogen atom.

(2)

Leave	
blank	

C	and to other mediations informal mediation of 1 122	
absorb	ared to other radiations, infrared radiation of wavelength 3.3 µm is stronged by hydrogen chloride gas. As a result of this absorption, the amplitudations of the hydrogen atoms significantly increases.	
(i) W	That name is given to this phenomenon?	
		•••••
		(1)
(ii) Sta	ate the condition for it to occur.	(1)
		•••••
(iii) Ca	alculate the frequency of infrared radiation of wavelength 3.3 μm.	(1)
	Frequency =	
(iv) II.		(2)
	ence calculate the stiffness of the hydrogen chloride bond.	
	ence calculate the stiffness of the hydrogen chloride bond.	
	ence calculate the stiffness of the hydrogen chloride bond.	
	ence calculate the stiffness of the hydrogen chloride bond.	
	ence calculate the stiffness of the hydrogen chloride bond.	
	ence calculate the stiffness of the hydrogen chloride bond.	

(Total 11 marks)

Leave	
blank	

(3)

*18. (a) A planet of mass m orbits a star of mass M. The radius of orbit is r. By considering the force required for circular motion in this situation, show that the period T of the orbit is given by $T^2 = \frac{4\pi^2 r^3}{GM}$.

.....

- (b) Measurements have shown that star HD70642 has a planet which orbits the star with a period of about 6 years. The radius of the orbit is about 3× the radius of the Earth's orbit around the Sun.
 - (i) Use the formula in (a) to determine a value for the ratio $\frac{\text{mass of star HD70642}}{\text{mass of Sun}}$

 $\frac{\text{mass of star HD70642}}{\text{mass of Sun}} = \dots$ (3)

(ii) Because of the presence of the planet, the star HD70642 does not remain at rest. Instead, the planet and star both orbit around their common centre of mass. Explain why the orbiting speed of the star is very small in comparison to the speed of the planet. (2) (c) Astronomers discovered the planet by observing the "Doppler Wobble" effect. As the planet orbits the star, light from the star undergoes a Doppler shift in its frequency. Explain why this method is likely to only detect very large planets.			Le
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	(c)	planet orbits the star, light from the star undergoes a Doppler shift in its frequence	
(Total 11 marks)			Q
		(Total 11 mark	(s)

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*19. Read the following passage and answer the questions that follow.

A nova is a sudden brightening of a star. Novae are thought to occur on the surface of a white dwarf star which is paired with another star in a binary system. If these two stars are close enough to each other, hydrogen can be pulled from the surface of the star onto the white dwarf. Occasionally, the temperature of this new material on the surface of the white dwarf may become hot enough for the hydrogen to fuse to helium. This causes the white dwarf to suddenly become very bright. In a nova, this hydrogen fusion occurs by the "CNO" process, where helium-4 is produced by a series of steps in which protons react with various isotopes of Carbon, Nitrogen and Oxygen. Novae are used by astronomers as standard candles.

(a) Complete the equation which shows a typical part of the CNO process.

$$^{14}_{8}O + ^{14}_{8}N + ^{14}_{8}N + ^{14}_{8}N$$
 (3)

(b)	What is a white dwarf? Suggest why hydrogen fusion in the white dwarf is likely t be the CNO process.	o
	(3	3)

	The	temperature required for these processes is 10^7 K.
		Calculate the mean kinetic energy, in keV, of the particles involved.
		energy of particles = \dots keV (3)
	(ii)	Explain how this temperature arises.
		(2)
d)	Astı	ronomers use novae as standard candles. Explain what a standard candle is, and
d)		ronomers use novae as standard candles. Explain what a standard candle is, and gest what this implies about the processes occurring in a nova.
d)		ronomers use novae as standard candles. Explain what a standard candle is, and
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